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510 CHAPTER 17: CHEMISTRY IN THE ATMOSPHERE The wavelength can now be calculated.  $3.48 \times 10^{-19} \text{ J}$   $(6.63 \times 10^{-34} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})$   $2.60 \times 10^{-10} \text{ m}$   $7.64 \times 10^{-10} \text{ m}$   $hc/E = \lambda$  17.12 Strategy: We are given the wavelength of the emitted photon and asked to calculate its energy. Equation (7.2) of the text relates the energy and frequency of an electromagnetic wave.

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The moon has no atmosphere and Venus has an atmosphere much denser than Earth's and atmospheric gases easily slow the speed of smaller incoming meteoroids. All of the dormant volcanoes on Venus and Mars are the shield type, and many are much larger than any shield volcano on Earth.

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### **HEATING THE ATMOSPHERE 17.2 NOTES**

Chapter 17 Atmosph The Atmosphere - Structure and Temperature. Question Answer; ozone: A molecule of oxygen containing 3 oxygen atoms - Oxygen molecules (2 atoms) absorb UV radiation and split into two separate atoms. These separate atoms collide with an oxygen molecule (2 atoms) and bond forming ozone (3 atoms).

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### **Chapters 4, 6, 7, 16, 17, 18 Environmental Systems**

BARRINGTON, Ill. (CBS) — Upset parents protested in Barrington Tuesday night over the district's plan to start the school year with remote learning. "I think this is child abuse, frankly ...

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